Silver Ion-Mediated Coupling of 4-Bromo-2,6-di-tert-butylcycohexa-2,5-dienone with Phenols

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Recently, we have shown that the Ag ion can induce nucleophilic displacement of 4-bromo-2,6-di-tert-butylcyclohexa-2,5-dienone (1) with alcohols.¹ We were then prompted to study the Ag ion-promoted displacement of 1 with phenolic nucleophiles. The products expected were 3,5-di-tert-butyl-4,4'- or 4,2'-biphenols which are not easily accessible by the conventional coupling reactions. This paper deals with the results of such a study.

Compound 1 was allowed to react at 0 °C for a short period with excess 2,6-diisopropylphenol (3) (4 mol equiv) in DME containing AgClO₄ (1 molar equiv). 3,5-Di-tertbutyl-3',5'-diisopropyl-4,4'-biphenol (4) was, as anticipated, obtained in 37% yield (eq 1).² The major product



was 4-bromo-2,6-di-tert-butylphenol (5) (47%). A small amount of 2,6-di-tert-butylphenol (8) (6%) was also produced. Prototropic rearrangement of 1 to give 5 was presumed to have been caused by catalysis with HClO₄, which was likely generated during the course of the reaction (eq 2, cf. eq 1). Neutralization of the HClO₄ as it was generated, therefore, was expected to suppress the



formation of 5. The AgClO₄-induced reaction of 1 with 3

in DME was carried out in the presence of excess Na₂- CO_3 (4 molar equiv), and the formation of 5 was found to be indeed minimized (3%), while that of 4 was substantially increased (60%).³ Symmetrically substituted 4,4'-biphenols 6a or 6b were not obtained. 4,4'-Diphenoquinones 7a, 7b, or 7c, the products of dehydrogenation of 6a, 6b, or 4, respectively, were not obtained either.



The success of preparation of **4** led us to investigate the coupling of **1** with other phenols by using $AgClO_4$ Na₂CO₃ in DME. The phenolic nucleophiles chosen were mainly alkyl-substituted ones. The results are shown in Table 1. From 4-unsubstituted 2,6-dialkylphenols 8, 9, 11, 13 and 15, 3,5-di-*tert*-butyl-3',5'-dialkyl-4,4'-biphenols 6b, 10, 12, 14, and 16, respectively, were generated in 50-68% yield. The coupling with 6-unsubstituted 2,4dialkylphenols 17, 20, and 22 afforded 3,5-di-tert-butyl-3',5'-dialkyl-4,2'-biphenol 19, 21, and 24, respectively, in 53-69% yield, along with the 3,5-di-tert-butyl-4-phenoxyphenols, the products of C-O coupling, as the minor products. From the reaction of 2,6-unsubstituted 4-alkylphenols 25 and 28, both the 4,2'-biphenols and the 4-phenoxyphenols were produced, the amounts of the latter products having been principal or substantial. All of the three possible coupling products, 34, 35, and 36, were obtained from the reaction of 3,5-dimethylphenol (33). In contrast, the reaction of 3,5-di-tert-butylphenol (31) afforded only 4-phenoxyphenol 32. Neither the 4,4'nor the 4,2'-biphenol was produced. Bulky tert-butyl groups at the 3- and 5-positions in **31** may prohibit attack by cation **2** on the ring carbons. While no or little C-Ocoupling took place in the reaction of 4-unsubstituted 2,6dimethylphenols 9, 13, and 15, the reaction of 2,4,6trimethylphenol (37) gave 4-phenoxyphenol 38, albeit in a fair vield.

The reaction of some phenols bearing a substituent(s) other than an alkyl group was also investigated. The reaction of 2,6-dimethoxyphenol (39) afforded 4,4'-biphenol 40. 4,3'-Biphenol 41 and 4,4'-diphenoquinones 42 and 7b were obtained in addition. The obtention of 41 as the principal C-C coupling product may probably indicate that the 3- and 5-positions in 39 are the most activated sites toward electrophilic attack by cation 2. Bromination of **39** with Br₂ has been reported to give 3-bromo-2,6dimethoxyphenol exclusively or almost exclusively rather than the 4-bromo isomer.⁴ The reaction of 4-chlorophenol (43), 4-acetylphenol (46), and 4-nitrophenol (49) gave both the expected 4-phenoxyphenols and the 4,2'-biphe-

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(1) Omura, K. J. Org. Chem. 1996, 61, 7156.
(2) A similar reaction with AgOCOCF₃, another Ag salt soluble in DME, in place of AgClO₄ provided only a few percent of 4. The ¹H NMR spectrum of the crude reaction product suggested that it contained significant amounts of 4-(trifluoroacetoxy)-2,6-di-tert-butylphenol and the termination of the crude reaction of 2.5 dill. tert butyley lohoya. 2.5 dill. its tautomer, 4- (trifluoroacetoxy)-2,6-di-tert-butylcyclohexa-2,5-dienone. Cf. ref 1.

⁽³⁾ The reaction with 2 molar equiv (in place of 4 mol equiv) of 3 gave 4 in 49% yield.

⁽⁴⁾ Giles, R. G.; Hughes, A. B.; Sargent, M. V. J. Chem. Soc., Perkin Trans. 1 1991, 1581



^a Not examined.

nols. However, the total yields of the coupling products obtained from the strongly deactivated phenols (**46** and **49**) were limited.

Of the biphenolic products listed in Table 1, symmetrically substituted 4,4'-biphenol **6b** can be prepared by dehydrogenative dimerization of $\mathbf{8}$.⁵ The other biphenols (except **41**) may in principle be producible by a oneelectron cooxidation of **8** and the appropriate phenols. The biphenols from this cross-coupling, however, may necessarily be accompanied by the biphenolic products from dehydrogenative dimerization of **8** and/or of the phenols employed. Moreover, further dehydrogenation of

^{(5) (}a) Ley, K.; Müller, E.; Mayer, R.; Scheffler, K. *Chem. Ber.* **1958**, *91*, 2670. (b) Kharasch, M. S.; Joshi, B. S. *J. Org. Chem.* **1957**, *22*, 1439. (c) Omura, K. *J. Org. Chem.* **1991**, *56*, 921. (d) Okada, K.; Okubo, K.; Oda, M.; Murai, H. *Chem. Lett.* **1995**, 845.

the biphenolic products to give the diphenoquinones may not be avoided. As a result, formation of the biphenols by the cross-coupling may not be very selective. For instance, cooxidation of 8 with 9, 15, or 39 with manganese complexes has been shown to give mixtures of products, from which 10, 16, or 40, respectively, are isolable in low yields.6

A convenient method for preparation of 3,5-di-tertbutyl-4-phenoxyphenols 53 may be the acid-catalyzed detert-butylation of 2,4,6-tri-tert-butyl-4-phenoxycyclohexa-2,5-dienones **52** (eq 3),⁷ which are obtained by reaction of 2,4,6-tri-tert-butylphenoxy radical with phenols.8 Indeed **32** has been prepared by using this method.⁷ It



should, however, be noted that the reaction to give 52 does not always proceed smoothly.9 In addition, selectivity of the formation of 53 from 52 is not always high.⁷

The results shown above may suffice to prove that 1 can couple with a wide variety of phenols when mediated by AgClO₄/Na₂CO₃. Exhaustive de-tert-butylation of the coupling products obtained here may be expected to furnish a number of new 4,4'- and 2,4'-biphenols as well as 4-phenoxyphenols.¹⁰

Experimental Section

¹H (90 MHz) and ¹³C (22.6 MHz) NMR and IR spectra were taken in CDCl₃ and in CHCl₃, respectively. Column chromatography was conducted on Merck SiO_2 60 using gradient elution (100% petroleum ether to 100% benzene). TLC was run on SiO₂. GC analyses were performed at 150 °C on a column packed with 10% FAP-S on Chromosorb W

General Procedure for Reactions in Table 1. To a mixture of powdered 4-bromo-2,6-di-tert-butylcyclohexa-2,5-dienone (1) (570 mg, 2 mmol) and finely powdered anhydrous Na₂-CO₃ (0.85 g, 8 mmol) was added in one portion a cold (0 °C) solution of a phenol (8 mmol) and $AgClO_4$ (415 mg, 2 mmol) in DME (10 mL), and the resulting mixture was stirred magnetically for 10 min at 0 °C. The mixture was filtered into a flask containing water. The contents of the flask were extracted with ether. The extract was washed with water, dried over anhydrous Na₂SO₄, and evaporated to dryness. The residue was chromatographed. The residues obtained from the reactions of 2,6dimethylphenol (9), 2,3,6-trimethylphenol (13), and 2,3,5,6tetramethylphenol (15) were sublimed (85 °C, 21 mmHg) to recover the unreacted phenols as the sublimates, and the residues were chromatographed. The residue obtained from the reaction of 2,6-dimethoxyphenol (39) was washed with petroleum ether, affording 3,5-di-tert-butyl-3',5'-dimethoxy-4,4'-diphenoquinone (42) as purple crystals (50 mg, 7%): mp 240-242 °C (lit.¹¹ mp 240-242 °C). The washings were evaporated, and the residue was chromatographed. Chromatography of the residue obtained from the reaction of 2,6-diisopropylphenol (3) provided, as the first fraction, a mixture of 2,6-di-tert-butylphenol (8) (35

(8) (a) Müller, E.; Ley, K.; Schlechte, G. *Chem. Ber.* 1957, *90*, 2660.
(b) Becker, H.-D. *J. Org. Chem.* 1964, *29*, 3068.
(9) Matsuura, T.; Nishinaga, A.; Ogura, K.; Omura, K. *J. Org. Chem.*

mg, 8%) and 2,6-di-tert-butyl-4-bromophenol (5)^{5a} (18 mg, 3%) as analyzed by GC and ¹H NMR spectroscopy.

3,5-Di-tert-butyl-3',5'-diisopropyl-4,4'-biphenol (4): 460 mg (60%) from 3; colorless crystals from hexane; mp 150-151.5 °C; ¹H NMR δ 7.31 (s, 2 H), 7.18 (s, 2 H), 5.16 (s, 1 H), 4.74 (s, 1 H), 3.20 (hept, J = 6.8 Hz, 2 H), 1.49 (s, 18 H), 1.32 (d, J = 7.0Hz, 12 H); $^{13}\!{\rm \hat{C}}$ NMR δ 152.7, 148.8, 135.8, 134.9, 133.5, 133.4, 123.7, 122.3, 34.4, 30.3, 27.4, 22.7; IR 3610, 3590 cm⁻¹. Anal. Calcd for C₂₆H₃₈O₂: C, 81.62; H, 10.01. Found: C, 81.37; H, 10.16

3,5,3',5'-Tetra-*tert*-butyl-4,4'-biphenol (6b): 465 mg (57%) from 8; light yellow crystals from hexane, identical with an authentic sample^{5c} (¹H NMR and TLC); mp 185–186.5 °C (lit.^{5c} mp 184.5–185.5 °C); ¹H NMR δ 7.30 (s, 4 H), 5.15 (s, 2 H), 1.49 (s, 18 H); ¹³C NMR δ 152.6, 135.8, 133.7, 123.9, 34.4, 30.3.

3,5-Di-tert-butyl-3',5'-dimethyl-4,4'-biphenol (10): 374 mg (57%) from 9; light yellow crystals from EtOH; mp 198.5-200 °C (lit.⁶ mp 198.5–200 °C). The ¹H NMR and IR spectra were in accordance with those reported for 10.6 13 C NMR δ 152.7, 151.0, 135.9, 134,5, 132.4, 127.1, 123.5, 123.0, 34.4, 30.4, 16.0. Anal. Calcd for C₂₂H₃₀O₂: C, 80.93; H, 9.26. Found: C, 80.85; H. 9.42

3,5,3'-Tri-tert-butyl-5'-methyl-4,4'-biphenol (12): 450 mg (61%) from 2-tert-butyl-6-methylphenol (11); colorless crystals from hexane; mp 177-178 °C; ¹H NMR & 7.30 (s, 2H), 7.26 (d, J = 2.3 Hz, 1 H), 7.13 (d, J = 2.3 Hz, 1 H), 5.15 (s, 1 H), 4.71 (s, 1 H), 2.29 (s, 3 H), 1.48 (s, 18 H), 1.46 (s, 9 H); 13 C NMR δ 152.6, $151.4,\,135.8,\,135.5,\,134.1,\,133.0,\,127.0,\,123.9,\,123.6,\,123.0,\,34.6,$ 34.4, 30.3, 29.8, 16.1; IR 3615, 3590 $\rm cm^{-1}.$ Anal. Calcd for C₂₅H₃₆O₂: C, 81.47, H, 9.85. Found: C, 81.32; H, 9.82

3,5-Di-tert-butyl-2',3',5'-trimethyl-4,4'-biphenol (14): 459 mg (68%) from 13; colorless crystals from hexane; mp 137.5-138 °C; ¹H NMR & 7.06 (s, 2 H), 6.90 (s, 1 H), 5.13 (s, 1 H), 4.59 (s, 1 H), 2.24 (s, 6 H), 2.16 (s, 3 H), 1.46 (s, 18 H); $^{13}\mathrm{C}$ NMR δ 152.1, 150.6, 135.4, 135.1, 133.4, 132.8, 129.3, 126.1, 122.1, 119.5, 34.4, 30.4, 17.5, 15.8, 12.3; IR 3615, 3590 cm⁻¹. Anal. Calcd for C23H32O2: C, 81.13; H, 9.47. Found: C, 80.93; H, 9.72.

3,5-Di-*tert*-butyl-2',3',5',6'-tetramethyl-4,4'-biphenol (16): 357 mg (50%) from 15; colorless crystals from hexane; mp 208-210 °C (lit.⁶ mp 199-200 °C). The ¹H NMR and IR spectra were in accordance with those reported for 16.6 $^{13}\mathrm{C}$ NMR δ 151.8, 150.3, 135.7, 135.4, 133.1, 133.1, 126.2, 118.8, 34.4, 30.6, 17.9, 12.3.

3,5-Di-tert-butyl-4-(2,4-di-tert-butylphenoxy)phenol (18): 76 mg (9%) from 2,4-di-*tert*-butylphenol (17); colorless oil; ¹H NMR δ 7.37 (d, J = 2.6 Hz, 1 H), 7.06 (dd, J = 2.4 Hz, 8.6 Hz, 1 H), 6.83 (s, 2 H), 6.59 (d, J = 8.6 Hz, 1 H), 4.94 (s, 1 H) 1.47 (s, 9 H), 1.41 (s, 18 H), 1.31 (s, 9 H); $^{13}\mathrm{C}$ NMR δ 155.0, 149.5, 148.8, 143.6, 137.9, 137.2, 123.7, 123.2, 116.7, 116.2, 34.9, 34.5, 34.3, 31.6, 30.2, 30.1; IR 3620 cm-1. HRMS calcd for C₂₈H₄₂O₂ 410.3185, found 410.3159.

3,5,3',5'-Tera-tert-butyl-4,2'-biphenol (19): 495 mg (60%) from 17; colorless crystals from MeOH, identical with an authentic sample12 (1H and 13C NMR, IR and TLC); mp 189-190.5 °C (lit.¹² mp 189-190.5 °C)

3,5,3'-Tri-tert-butyl-5'-methyl-4,2'-biphenol (21): 511 mg (69%) from 2-tert-butyl-4-methylphenol (20); colorless crystals from hexane; mp 137.5–139 °Č; ¹H NMR δ 7.20 (s, 2 H), 7.06 (d, J = 2.3 Hz, 1 H), 6.9–6.8 (m, 1 H), 5.41 (s, 1 H), 5.28 (s, 1 H), 2.30 (s, 3 H), 1.46 (s, 18 H), 1.44 (s, 9 H); $^{13}\mathrm{C}$ NMR δ 153.4, 148.8, 136.8, 135.4, 129.3, 128.3, 128.2, 128.0, 126.6, 126.0, 34.8, 34.5, 30.3, 29.7, 20.8; IR 3615, 3525 cm⁻¹. Anal. Calcd for C25H36O2: C, 81.47; H, 9.85. Found: C,81.32; H, 9.94

3,5-Di-tert-butyl-4-(2,4-dimethylphenoxy)phenol (23): 77 mg (12%) from 2,4-dimethylphenol (22); light yellow oil; ¹H NMR δ 7.1-6.6 (m, 3 H), 6.79 (s, 2 H), 4.89 (s, 1 H), 2.28 (s, 3 H), 2.25 (s, 3 H), 1.40 (s, 18 H); ¹³C NMR δ 153.7, 149.5, 149.1, 137.2, 131.6, 131.5, 127.0, 116.9, 115.1, 34.5, 30.2, 20.6, 16.2; IR 3620 cm⁻¹; HRMS calcd for $C_{22}H_{30}O_2$ 326.2247, found 326.2238.

3,5-Di-tert-butyl-3',5'-dimethyl-4,2'-biphenol (24): 346 mg (53%) from 22; colorless crystals from hexane; mp 167.5-169 °C; ¹H NMR δ 7.21 (s, 2 H), 7.0–6.8 (m, 2 H), 5.28 (s, 1 H), 5.20 (s, 1 H), 2.27 (s, 6 H), 1.46 (s, 18 H); ¹³C NMR δ 153.4, 148.3, 136.7, 130.4, 128.8, 128.1, 127.9, 125.6, 123.9, 34.5, 30.3, 20.4,

⁽⁶⁾ Nishino, H.; Itoh, N.; Nagashima, M.; Kurosawa, K. Bull. Chem. Soc. Jpn. 1992, 65, 620.

⁽⁷⁾ Rieker, A.; Zeller, N. Terahedron Lett. 1968, 4969.

^{1969,} *34*, 550.

^{(10) (}a) Olah, G. A.; Surya Prakash, G. K.; Iyer, P. S.; Tashiro, M.; Yamato, T. J. Org. Chem. **1987**, 52, 1881. (b) Tashiro, M.; Yoshiya, H.; Yamato, T. Synthesis **1978**, 399.

⁽¹¹⁾ Boldt, P.; Michaelis, W.; Lackner, H.; Krebs, B. Chem. Ber. 1971, 104, 220.

⁽¹²⁾ Omura, K. Terahedron 1995, 51, 6901.

16.1; IR 3620, 3535 cm $^{-1}$ Anal. Calcd for $C_{22}H_{30}O_2:\,\,C,80.94,\,H,$ 9.26. Found: C, 80.92; H, 9.44.

3,5-Di-*tert*-butyl-4-(4-*tert*-butylphenoxy)phenol (**26**): 472 mg (67%) from 4-*tert*-butylphenol (**25**); colorless crystals from hexane; mp 139–140 °C; ¹H NMR δ 7.4–7.1 (m, 2 H), 7.0–6.7 (m, 2 H), 6.88 (s, 2 H), 4.97 (s, 1 H), 1.41 (s, 18 H), 1.31 (s, 9 H); ¹³C NMR δ 156.3, 149.8, 148.4, 144.3, 137.2, 126.1, 116.6, 116.1, 34.5, 34.1, 31.5, 30.2; IR 3620 cm⁻¹. Anal. Calcd for C₂₄H₃₄O₂: C, 81.31; H, 9.67. Found: C, 81.02; H, 9.51.

3,5,5'-Tri-*tert***-butyl-4,2'-biphenol (27):** 190 mg (27%) from **25**; colorless crystals from hexane; mp 136–137 °C; ¹H NMR δ 7.4–7.1 (m, 2 H), 7.24 (s, 2 H), 7.0–6.8 (m, 1 H), 5.30 (s, 1 H), 5.15 (s, 1 H), 1.47 (s, 18 H), 1.33 (s, 9 H); ¹³C NMR δ 153.4, 150.1, 143.1, 136.7, 128.2, 128.1, 126.9, 125.7, 125.3, 114.8, 34.5, 34.1, 31.6, 30.3; IR 3620, 3540, 3300 cm⁻¹. Anal. Calcd for C₂₄H₃₄O₂; C, 81.31; H, 9.67. Found: C, 81.05; H, 9.86.

3,5-Di-*tert***-butyl-4-(4-methylphenoxy)phenol (29):** 188 mg (30%) from 4-methylphenol (**28**); colorless crystals from hexane; mp 98–99 °C; ¹H NMR δ 7.2–7.0 (m, 2 H), 6.9–6.6 (m, 2 H), 6.86 (s, 2 H), 4.95 (s, 1 H), 2.30 (s, 3 H), 1.41 (s, 18 H); ¹³C NMR δ 156.2, 149.6, 148.9, 137.2, 131.2, 129.8, 117.1, 116.1, 34.5, 30.2, 20.6; IR 3615 cm⁻¹. Anal. Calcd for C₂₁H₂₈O₂: C, 80.73; H, 9.03. Found: C, 80.77; H, 9.00.

3,5-Di-*tert***-butyl-5**′-**methyl-4**,2′-**biphenol (30):** 245 mg (39%) from **28**; colorless crystals from hexane; mp 50-54 °C; ¹H NMR δ 7.22 (s, 2 H), 7.1–6.8 (m, 3 H), 5.29 (s, 1 H), 5.14 (s, 1 H), 2.31 (s, 3 H), 1.47 (s, 18 H); ¹³C NMR δ 153.4, 150.2, 136.7, 130.5, 129.5, 128.8, 128.5, 127.8, 125.5, 115.1, 34.5, 30.3, 20.5; IR 3620, 3520, 3300 cm⁻¹. Anal. Calcd for C₂₁H₂₈O₂: C, 80.73; H, 9.03. Found: C, 80.66; H, 9.15.

3,5-Di-*tert***-butyl-4-(3,5-di-***tert***-butylphenoxy)phenol (32):** 638 mg (78%) from 3,5-di-*tert*-butylphenol (**31**); colorless crystals from hexane; mp 119.5–121 °C (lit.⁷ mp 112–114 °C); ¹H NMR δ 7.09 (t, J = 1.7 Hz, 1 H), 6.89 (s, 2 H), 6.82 (d, J = 1.7 Hz, 2 H), 4.93 (s, 1 H), 1.41 (s, 18 H), 1.29 (s, 18 H); ¹³C NMR δ 157.3, 152.1, 149.2, 148.8, 137.1, 116.0, 115.6, 111.8, 34.9, 34.5, 31.4, 30.2; IR 3620 cm⁻¹.

3,5-Di-*tert***-butyl-4-(3,5-dimethylphenoxy)phenol (34):** 127 mg (19%) from 3,5-dimethylphenol (33); colorless crystals from hexane; mp 54–55 °C; ¹H NMR δ 6.87 (s, 2 H), 6.65 (s, 1 H), 6.56 (s, 2 H), 4.97 (s, 1 H), 2.26 (s, 6 H), 1.41 (s, 18 H); ¹³C NMR δ 158.5, 149.7, 148.5, 139.1, 137.2, 123.5, 116.4, 114.7, 34.5, 30.2, 21.3; IR 3620 cm⁻¹. Anal. Calcd for C₂₂H₃₀O₂: C, 80.93; H, 9.26. Found: C, 80.96, H, 9.16.

3,5-Di-*tert***-butyl-2'**,**6'**-**dimethyl-4,4'**-**biphenol (35)** was obtained as a mixture with **33** (starting phenol) upon chromatography. A solution of the mixture in diisopropyl ether was shaken with aqueous NaOH. The organic layer was washed with water, dried, and evaporated to leave **35** (152 mg, 23%). Colorless crystals from diisopropyl ether/hexane: mp 214–240 °C; ¹H NMR δ 6.88 (s, 2 H), 6.58 (s, 2 H), 5.11 (s, 1 H), 4.63 (s, 1 H), 2.01 (s, 6 H), 1.44 (s, 18 H); ¹³C NMR δ 153.6, 152.0, 138.1, 135.5, 131.2, 125.9, 113.8, 34.4, 30.5, 21.1; IR 3620, 3585, 3325 cm⁻¹. Anal. Calcd for C₂₂H₃₀O₂: C, 80.93; H, 9.26. Found: C, 80.73; H, 9.48.

3,5-Di-*tert*-**butyl**-**4'**,**6'**-**dimethyl**-**4**,**2'**-**biphenol** (**36**): 292 mg (45%) from **33**; colorless crystals from hexane; mp 171–172 °C; ¹H NMR δ 7.04 (s, 2 H), 6.68 (s, 2 H), 5.28 (s, 1 H), 4.92 (s, 1 H), 2.31 (s, 3 H), 2.05 (s, 3 H), 1.45 (s, 18 H); ¹³C NMR δ 153.4, 152.9, 137.9, 137.1, 136.8, 126.7, 125.9, 125.5, 122.5, 112.8, 34.5, 30.3, 21.2, 20.4; IR 3620, 3520, 3300 cm⁻¹. Anal. Calcd for C₂₂H₃₀O₂: C, 80.93; H, 9.26. Found: C, 80.83; H, 9.43.

3,5-Di-*tert*-butyl-4-(2,4,6-trimethylphenoxy)phenol (38): 236 mg (35%) from 2,4,6-trimethylphenol (37); colorless crystals from hexane; mp 104–105 °C; ¹H NMR δ 6.85 (s, 2 H), 6.57 (s, 2 H), 4.71 (s, 1 H), 2.28 (s, 3 H), 2.08 (s, 6 H), 1.36 (s, 18 H); ¹³C NMR δ 150.8, 149.1, 147.5, 137.0, 133.5, 130.9, 129.2, 110.8, 34.5, 30.2, 20.8, 16.4; IR 3625 cm⁻¹. Anal. Calcd for C₂₃H₃₂O₂: C, 81.13; H, 9.47. Found: C, 80.98; H, 9.51.

3,5-Di-*tert***-butyl-3',5'-dimethoxy-4,4'-biphenol (40):** 156 mg (22%) from **39**; light yellow crystals from diisopropyl ether, identical with an authentic sample (¹H NMR, IR, and TLC)

kindly provided by Dr. Nishino; mp 145.5–147 °C (lit. 6,13 mp 201–201.5 °C). 13 C NMR δ 153.0, 147.0, 136.1, 134.0, 133.8, 132.9, 123.7, 104.2, 56.4, 34.4, 30.3.

3,5-Di-*tert*-**butyl**-**2**',**4**'-**dimethoxy**-**4**,**3**'-**biphenol (41)**: 282 mg (39%) from **39**; light yellow oil; ¹H NMR δ 7.36 (s, 2 H), 6.83 (AB, J = 8.6 Hz, 1 H), 6.69 (AB, J = 8.6 Hz, 1 H), 5.75 (s, 1 H), 5.19 (s, 1 H), 3.89 (s, 3 H), 3.60 (s, 3 H), 1.47 (s, 18 H); ¹³C NMR δ 152.7, 146.4, 144.5, 138.6, 135.4, 128.9, 128.6, 125.4, 120.1, 106.8, 60.2, 56.2, 34.4, 30.4; IR 3620, 3520 cm⁻¹; HRMS calcd for C₂₂H₃₀O₄ 358.2145, found 358.2147.

3,5,3',5'-Tetra-*tert***-butyl-4,4'-diphenoquinone (7b):** 36 mg (9%) from the reaction of **39**; reddish brown crystals from benzene, identical with an authentic sample¹⁴ (¹H NMR and TLC); mp 246–248 °C (lit.¹⁴ mp 240–241 °C).

3,5-Di-*tert*-butyl-4-(4-chlorophenoxy)phenol (44): 379 mg (57%) from 4-chlorophenol (43); colorless crystals from hexane; mp 82.5–83.5 °C; ¹H NMR δ 7.3–7.1 (m, 2 H), 6.9–6.7 (m, 2 H), 6.86 (s, 2 H), 5.01 (s, 1 H), 1.41 (s, 18 H); ¹³C NMR δ 157.3, 150.1, 148.2, 137.5, 129.2, 126.7, 118.2, 116.4, 34.5, 30.2; IR 3615 cm⁻¹. Anal. Calcd for C₂₀H₂₅O₂Cl; C, 72.17; H, 7.57; Cl, 10.65. Found: C, 72.30; H, 7.73; Cl, 10.45.

3,5-Di-*tert*-butyl-5'-chloro-4,2'-biphenol (45): 152 mg (23%) from 43; colorless crystals from hexane; mp 107–108 °C; ¹H NMR δ 7.3–7.1 (m, 2 H), 7.19 (s, 2 H), 7.0–6.8 (m, 1 H), 5.34 (s, 1 H), 5.27 (s, 1 H), 1.46 (s, 18 H); ¹³C NMR δ 153.9, 151.1, 137.0, 130.3, 129.6, 128.1, 126.4, 125.4, 125.1, 116.6, 34.5, 30.2; IR 3620, 3530, 3300 cm⁻¹. Calcd for C₂₀H₂₅O₂Cl: C, 72.17; H, 7.57; Cl, 10.65. Found: C, 72.04; H, 7.62; Cl, 10.72.

3,5-Di-*tert***-butyl-4-(4-acetylphenoxy)phenol (47):** 158 mg (23%) from 4-acetylphenol (**46**); colorless crystals from diisopropyl ether; mp 128–130 °C; ¹H NMR δ 8.0–7.8 (m, 2 H), 7.0–6.8 (m, 2 H), 6.89 (s, 2 H), 5.09 (s, 1 H), 2.55 (s, 3 H), 1.43 (s, 18 H); ¹³C NMR δ 196.4, 163.0, 150.6, 147.1, 137.7, 131.0, 130.4, 117.0, 116.0, 34.5, 30.2, 26.3; IR 3620, 1673 cm⁻¹. Anal. Calcd for C₂₂H₂₈O₃: C, 77.61; H, 8.29. Found: C, 77.51; H, 8.47.

3,5-Di-*tert*-butyl-5'-acetyl-4,2'-biphenol (48): 121 mg (18%) from 46; colorless crystals from MeOH/hexane; mp 169–170 °C; ¹H NMR δ 8.0–7.8 (m, 2 H), 7.22 (s, 2 H), 7.1–6.9 (m, 1 H), 5.98 (s, 1 H), 5.37 (s, 1 H), 2.57 (s, 3 H), 1.47 (s, 18 H); ¹³C NMR δ 196.8, 157.0, 153.9, 137.1, 131.0, 130.2, 129.4, 129.0, 126.5, 125.6, 115.3, 34.5, 30.2, 26.3; IR 3620, 3515, 3250, 1668 cm⁻¹. Anal. Calcd for C₂₂H₂₈O₃: C, 77.61; H, 8.29. Found: C, 77.53; H, 8.43.

3,5-Di-*tert***-butyl-4-(4-nitrophenoxy)phenol (50):** 217 mg (32%) from 4-nitrophenol (**49**); light yellow crystals from hexane; mp 131–132 °C; ¹H NMR δ 8.2–8.0 (m. 2 H), 7.0–6.8 (m, 2 H), 6.90 (s, 2 H), 5.15 (s, 1 H), 1.44 (s, 18 H); ¹³C NMR δ 164.2, 151.0, 146.6, 141.9, 137.9, 125.7, 117.1, 116.0, 34.5, 30.1; IR 3615 cm⁻¹. Anal. Calcd for C₂₀H₂₅O₄N: C, 69.95; H, 7.34; N, 4.08. Found: C, 69.88; H, 7.31; N, 3.79.

3,5-Di-*tert*-butyl-5'-nitro-4,2'-biphenol (51): 28 mg (4%) from 49; light yellow crystals from hexane; mp 135–138 °C; ¹H NMR δ 8.2–8.0 (m, 2 H), 7.20 (s, 2 H), 7.1–6.9 (m, 1 H), 6.00 (s, 1 H), 5.43 (s, 1 H), 1.48 (s, 18 H); ¹³C NMR δ 158.1, 154.4, 141.4, 137.4, 129.5, 126.0, 125.4, 125.1, 124.5, 115.7, 34.5, 30.2; IR 3620, 3515, 3200 cm⁻¹. Anal. Calcd for C₂₀H₂₅O₄N: C, 69.95; H, 7.34; N, 4.08. Found: C, 70.00; H, 7.33; N, 4.13.

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⁽¹³⁾ Dr. Nishino corrected the melting point to 148.5–149.0 $^{\circ}\mathrm{C}$ (private communication)

⁽¹⁴⁾ Omura, K. J. Org. Chem. 1984, 49, 3046.